

# Chemistry Suggestion Paper - Chemical Bond

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**1. Give definition with their properties:**

- Ionic bond / Electrovalent bond.
- Covalent bond.
- Co-ordinate Bond
- Metallic bond → (Electrical and thermal conductical nature, Luster, Ductility)
- Hydrogen bond → (definition, variation and properties of Dipole)
- Vander wall's bond
- Hybridization

**2. Explain the Ionic character fo Covalent bond.**

**3. Explain the Covalent character fo Ionic bond.**

**4. Fazan's rule**

**5. Write down the bonding angles of  $sp^3$  hybridized compound.**

**6. Explain these terms below with example:**

- Why  $NH_3$  is a pyramidal structure compound.
- Why  $H_2O$  is a 'V' shape structure compound.
- $sp^3$  hybridization for  $CH_4$
- $sp^2$  hybridization for  $CH_2 = CH_2$
- $sp$  hybridization for  $CH_3 \equiv CH_3$

**7. Why  $PCl_5$  is possible and  $NCl_5$  is a impossible compound.**

**8. Why  $H_2O$  is liquid and  $H_2S$  is gas in natural temperature.**

**9. Explain these compounds solibility in water:  $AgF$ ,  $AgCl$ ,  $AgBr$ ,  $AsI$**